Caustic soda Sodium hydrate Soda lye

MATERIAL SAFETY DATA SHEET

CAS No: 1310-73-2

RTECS No: WB4900000

UN No: 1823

EC No: 011-002-00-6

NaOH

Molecular mass: 40.0

TYPES OF HAZARD / EXPOSURE	ACUTE HAZARDS / SYMPTOMS	PREVENTION	FIRST AID / FIRE FIGHTING
FIRE	Not combustible. Contact with moisture or water may generate sufficient heat to ignite combustible substances.		In case of fire in the surroundings: all extinguishing agents allowed.
EXPLOSION			
EXPOSURE		AVOID ALL CONTACT!	IN ALL CASES CONSULT A DOCTOR!
Inhalation	Corrosive. Burning sensation. Sore throat. Cough. Laboured breathing. Shortness of breath. Symptoms may be delayed (see Notes).	Local exhaust or breathing protection.	Fresh air, rest. Half-upright position. Artificial respiration if indicated. Refer for medical attention.
Skin	Corrosive. Redness. Pain. Serious skin burns. Blisters.	Protective gloves. Protective clothing.	Remove contaminated clothes. Rinse skin with plenty of water or shower. Refer for medical attention.
Eyes	Corrosive. Redness. Pain. Blurred vision.	Face shield, or eye protection in	First rinse with plenty of water for several

	Severe deep burns.	combination with breathing protection if powder.	minutes (remove contact lenses if easily possible), then take to a doctor.
Ingestion	Corrosive. Burning sensation. Abdominal pain. Shock or collapse.	Do not eat, drink, or smoke during work.	Rinse mouth. Do NOT induce vomiting. Give plenty of water to drink. Refer for medical attention.

SPILLAGE DISPOSAL	PACKAGING & LABELLING	
Sweep spilled substance into suitable containers. Wash away remainder with plenty of water. (Extra personal protection: complete protective clothing including self-contained breathing apparatus).	© Symbol R: <u>35</u> S: (<u>1/2-)26-37/39-45</u> UN Hazard Class: 8 UN Pack Group: II	Unbreakable packaging; put breakable packaging into closed unbreakable container. Do not transport with food and feedstuffs.

EMERGENCY RESPONSE	STORAGE
Transport Emergency Card: TEC (R)-121 NFPA Code: H 3; F 0; R 1	Separated from strong acids, metals, food and feedstuffs. Dry. Well closed. Store in an area having corrosion resistant concrete floor.

IMPORTANT DATA

Physical State; Appearance

WHITE, DELIQUESCENT SOLID IN VARIOUS FORMS, WITH NO ODOUR.

Chemical dangers

The substance is a strong base, it reacts violently with acid and is corrosive in moist air to metals like zinc, aluminium, tin and lead forming a combustible/explosive gas (hydrogen - see ICSC 0001). Reacts with ammonium salts

to produce ammonia, causing fire hazard. Attacks some forms of

Routes of exposure

The substance can be absorbed into the body by inhalation of its aerosol and by ingestion.

Inhalation risk

Evaporation at 20°C is negligible; a harmful concentration of airborne particles can, however, be reached quickly.

Effects of short-term exposure

Corrosive. The substance is very corrosive to the eyes, the skin and the

plastics, rubber or coatings. Rapidly absorbs carbon dioxide and water from air. Contact with moisture or water may generate heat (see Notes).

respiratory tract. Corrosive on ingestion. Inhalation of an aerosol of the substance may cause lung oedema (see Notes).

Occupational exposure limits

TLV: 2 mg/m³ (ceiling values) (ACGIH 2000). MAK not established.

Effects of long-term or repeated exposure

Repeated or prolonged contact with skin may cause dermatitis.

PHYSICAL PROPERTIES	ENVIRONMENTAL DATA
Boiling point: 1390°C Melting point: 318°C Density: 2.1 g/cm³ Solubility in water, g/100 ml at 20°C: 109	This substance may be hazardous to the environment; special attention should be given to water organisms.

NOTES

The applying occupational exposure limit value should not be exceeded during any part of the working exposure.

The symptoms of lung oedema often do not become manifest until a few hours have passed and they are aggravated by physical effort. Rest and medical observation are therefore essential.

NEVER pour water into this substance; when dissolving or diluting always add it slowly to the water.

Other UN number: UN1824 Sodium hydroxide solution, Hazard class 8.

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