## **MATERIAL SAFETY DATA SHEET**

#### 1. CHEMICAL PRODUCT IDENTIFICATION

| Chemical Name    | : | Sodium Hypochlorite Solution<br>UN.No.1791                               |
|------------------|---|--|
| Synonym          | : | Bleach Hypochlorous acid, Sodium Salt, Soda bleach & Sodium Oxychloride. |
| Chemical Formula | : | Naocl  |
| Molecular weight | : | 74.44  |
| Product codes    | : | 9416, P005   |

#### 2. PHYSICAL AND CHEMICAL PROPERTIES

| Appearance    | : | Colourless to yellowish liquid |
|---------------|---|--------------------------------|
| Odor          | : | Chlorine like odor             |
| Solubility    | : | 100% in water                  |
| Density       | : | 1.07 - 1.14                    |
| Boiler point  | : | 40C(104F) Decomposer slightly  |
| Meeting point | : | -6C (21F)                      |

#### 3. STABILITY AND REACTIVITY

Stability : Slowly decomposes on contact with air. Rate increases with the concentration and temperature. Exposure to sunlight acceleration decomposition. Sodium Hypochlorite becomes less toxic with age.

Hazardous Decomposition Product:

Emits toxic fumes of Chlorine when heated to decomposition. Sodium oxide at high temperatures.

Hazardous Polymerization : Will not occur.

Continuous to avoid : Light, heat and incompatibles.

#### 4. HAZARDS IDENTIFICATION

Potential Health effects:

| Inhalation   | : | May cause irritation to the respiratory that (Nose        |
|--------------|---|---|
|              |   | and throat; symptoms may include coughing and             |
|              |   | sore throat)  |
| Ingestion    | : | May cause nausea and vomiting.                            |
| Skin contact | : | May irritate skin.  |
| Eye contact  | : | Contact may cause severe irritation and damage especially |

|   | at higher concentration.                                     |  |
|---|--|--|
| Chronic Exposure :                      | A constant irritant to the eyes and throat. Low potential    |  |
| -                                       | for sensitization after exaggerated exposure to damage skin. |  |
| Aggravation of pre-existing conditions: |  |  |
|   | Persons with impaired respiratory function, or heart         |  |
|   | disorders (or disease) may be more susceptible to the        |  |
|   | effects of substance.  |  |

#### 5. **<u>FIST AID MEASURES (Safety Precaution)</u>**:

| Inhalation    | : | Remove to fresh air. If not breathing, give artificial     |
|---------------|---|--|
|               |   | respiration. If breathing is difficult, give Oxygen, get   |
|               |   | medical attention immediately.                             |
| Ingestion     | : | If swallowed, DO NOT INDUCE VOMITING. Give large           |
|               |   | quantities of water. Never give anything by mouth to an    |
|               |   | unconscious person. Get medical attention immediately.     |
| Skin contact  | : | Immediately flush skin with plenty of water for atleast 15 |
|               |   | minutes while removing contaminated clothing and shoes.    |
|               |   | Get medical attention immediately. Wash clothing before    |
|               |   | re-use. Thoroughly clean shoes before re-use.              |
| Eye contact : | : | Immediately flush eyes with plenty of water for atleast 15 |
|               |   | minutes, lifting lower and upper eyelids occasionally. Get |
|               |   | medical attention immediately.                             |
|               |   |  |

### 6. **<u>FIRE FIGHTING MEASURES</u>**:

| Fire :                    | Not considered to be a fire hazard, substance releases<br>oxygen when heated, which may increase the severity of an<br>existing fire containers may rupture from pressure build-up.                    |
|---------------------------|--|
| Explosion :               | This solution is not considered to be an explosion hazard.<br>Anhydrous sodium hypochlorite is very explosive.   |
| Fire Extinguishing media: | Use any means suitable for extinguishing surrounding fire.<br>Use water spray to cool fire-exposed containers to dilute<br>liquid and control vapour.  |
| Special information :     | In the event of a fire, wear full protective clothing and NIOSH – approved self – contained breathing apparatus with full face shield operated in the pressure demand or other positive pressure mode. |

# Chemistry for every day Living